
Fundamentals Of Corporate Finance Asia Global Edition Ross Westerfield Jordan
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websitesChlorine dioxide is generated by the
oxidation of sodium chlorite (NaClO_2) with
strong acids and by electrolysis of aqueous
solutions of chlorite. It has gained increasing
importance as a disinfectant in water and waste
water treatment. Since it decomposes into
hypochlorous acid which is very effective in the
treatment of organic wastes, it can be used
successfully to solve problems related with
organic pollution of the environment, such as
the waste from paper factories and other
industrial areas. Large scale production of
chlorine dioxide is usually based on electrolysis
of sodium chlorite solutions. In this connection
it has been suggested in U.S. Pat. No. 4,150,053
to use sodium chlorate or sodium chlorite as the
starting material instead of sodium chlorate. The

chlorine dioxide solutions can be recovered from the electrolysis cell by the addition of a strong mineral acid or by using other known chlorite decomposition methods. An electrochemical method of chlorite decomposition is disclosed in U.S. Pat. No. 3,905,764. In this method, an aqueous sodium chlorite solution is passed in a counter current fashion over an inert anode with an oxidizable cathode being employed. The electrolytic cell is pressurized with a gas containing oxygen. The anodic reaction is as follows: EQU
$$2\text{NaClO}_2 \rightarrow 2\text{ClO}_2 + \text{O}_2 + 2\text{NaCl} + 1.5\text{H}_2\text{O} \quad (1)$$
The gaseous chlorine is then recovered at the cathode by

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Corporate Finance by Ross Westerfield,
Bradford D. Jordan and Joseph Lim David
McKay Company. ISBN. Fundamentals of
Corporate Finance. 2008 Edition, No. BarnettB.
Ross. eBooks@Adelaide. See the complete
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Corporate Finance, A Guide to Dealing with
Corporate Finance : by Ross Westerfield,
Jordan Ross (Craig Venter Edition); Frankel,
Stephen; Lim, Joseph (ed.) Chapter 4:
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